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# Determination of Volatile Organic Compounds (VOCs)

Mr. Tim Wong Accreditation Officer Hong Kong Accreditation Service

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- Sampling and Analytical Methods
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#### **Definitions**

- Hong Kong Air Pollution Control (VOCs) regulation (CAP 311W):
- VOC means any volatile compound of carbon excluding methane, carbon monoxide, carbon dioxide, carbonic acid, metallic carbides, metallic carbonates, ammonium carbonate and exempt compounds

- World Health Organisation WHO (1989):
- "any organic compounds"
- Very volatile organic compound (VVOC) <0 up to 50°C
- volatile organic compound (VOC) 50 100 up to 240 260°C
- semivolatile organic compound (SVOC) 240 260 up to 380 400°C

- United Nations Economic Commission for Europe UN-ECE protocol (nov. 1991):
- "..any organic compounds of anthropogenic nature, other than methane, that are capable of producing photochemical oxidants by reacting with nitrogen oxides in the presence of sunlight."
- This definition relates to ozone formation

- US Environmental Protection Agency US EPA:
- "any volatile compound of carbon excluding methane, carbon monoxide, carbon dioxide, carbonic acid, metallic carbides, metallic carbonates, ammonium carbonate and other compounds"
- This definition relates to ozone formation

- European Unit EU 2001/81/EC (National Ceiling Directive):
- "..VOC means all organic compounds arising from human activity, other than methane, which are capable of producing photochemical oxidants by reactions with nitrogenic oxides in the presence of sunlight."
- This definition relates to ozone formation

#### Sources

#### Potential Sources of VOC Emissions

- Ambient mobile and stationary sources (e.g. benzene from automotive exhaust and vapor release from gasoline service stations);
- Off-gassing of chemicals from furniture (e.g., formaldehyde from desks made of particleboard);
- Construction, demolition and building renovation activities (e.g., painting the walls);
- Personal hygiene products (e.g., perfumes and hairsprays);
- Pesticides and insecticides;
- Commercial activities (e.g., automotive painting and dry cleaning exhausts)

#### Hazards

#### Health Effects of VOC

- Carcinogenic Effects
  - The ability of a substance to cause cancer in humans and animals
  - Carcinogenicity of VOCs varies between species
  - According to US EPA, VOCs typically account for 35 to 55% of outdoor air cancer risk in the United States

#### Health Effects of VOC

- Non-Carcinogenic Effects
  - A number of VOCs if above certain endpoints tend to affect the respiratory, digestive and certain nervous systems
  - These endpoints vary widely between compounds and exposure pathways

#### Health Effects of VOC

- Odour and Annoyance
  - The malodorous properties of certain VOCs may lead to irritation or annoyance in humans
  - Odour threshold is used to describe the theoretical minimum concentration of odourant necessary to be detected
  - Odour threshold concentrations for many VOCs may be below those considered toxic to humans

#### **Environmental Effects of VOC**

- Ozone Formation Capacity
  - The formation of ground-level ozone (O<sub>3</sub>) through photochemical oxidation of certain VOCs
  - Negative impacts on human health, including lung damage, chest pains, coughing, nausea, throat irritation and congestion
  - Negative vegetative effects, such as the reduced ability of plants to produce and store food, making them more susceptible to disease, insects, other pollutants and harsh weather
  - Ozone is the primary component of smog, the brownish haze commonly seen over urban areas during the summer

#### **Environmental Effects of VOC**

- Particular Matter
  - Particulate matter (PM) includes solid particles as well as liquid droplets that are suspended in air
  - The most common size categories are: PM<sub>2.5</sub> and PM<sub>10</sub>
  - Elevated concentrations of PM<sub>10</sub> most commonly a result of direct natural emissions, primarily of materials derived from the earth's crust, such as soil and minerals
  - Elevated concentrations of PM<sub>2.5</sub> caused partly by direct emissions and partly by indirectly formation from pollutants such SO<sub>2</sub>, NO<sub>x</sub>, VOCs and NH<sub>3</sub>, therefore usually the result of anthropogenic activities

#### Examples

- Benzene
  - Boiling point: 80.1 °C
  - Second-hand tobacco smoking, solvents, paints, fax machines
  - Known to cause cancer
- Tetrachloroethylene
  - Boiling point: 121.1 °C
  - Dry-cleaned fabrics, furniture coverings, spot/textile cleaners
  - Suspected carcinogen

#### Examples

- Toluene
  - Boiling point: 111 °C
  - Solvent, perfumes, detergents, dyes, water-based adhesives
  - Insufficient data on cancer (animals)
- Trichlorofluoromethane
  - Boiling point: 23.8 °C
  - Refrigerants, fire extinguishers
  - Ozone depleting chemicals (banned)

### Sampling and Analytical Methods

#### Sample Matrices for VOCs Analysis

- Air
  - Indoor and Outdoor Air, Mobile and Stationary Sources
- Water
  - Wastewater, Drinking Water
- Soil/Sediment
- Products
  - Wooden Board, Hairspray

Types of Sampling Methods - Air

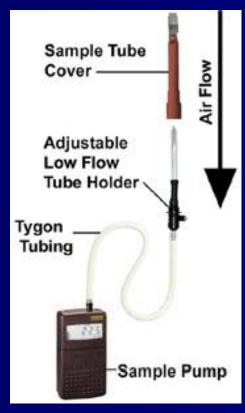
- Active sampling
  - The process of connecting sample collection media to a sampling train consisting of inert tubing and a sample pump operating at a know flow rate
  - The sampling pump draws air through the sample media, resulting in a know sample volume

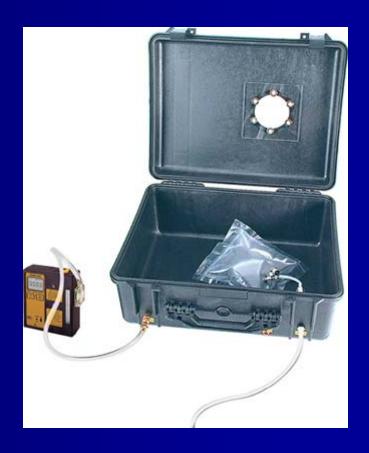
#### Types of Sampling Methods

- Active sampling
  - Inlet
  - Sampling Media (Tubes or Bags)
  - Tubing
  - Pump



Active sampling





\*Courtesy of SKC Corporation

■ Active sampling – Pumps







\*Courtesy of SKC Corporation

Active sampling – Pumps (Cont'd)





Active sampling – Sampling Tubes and Bags





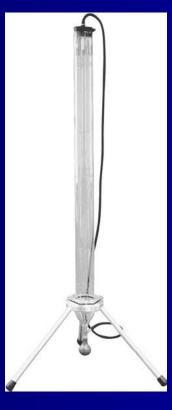
Active sampling – Calibration of Flow Rate





Active sampling – Calibration of Flow Rate (Cont'd)







Active sampling – Flow Rate Measurements





#### Types of Sampling Methods

- Passive sampling
  - allowing sample collection media (e.g., solvent or thermal desorption tube) to passively diffuse air through the sample media without benefit of 'forced air'.
  - This allows for longer potential sampling periods without concern for overloading the sampling media.

Passive samplers







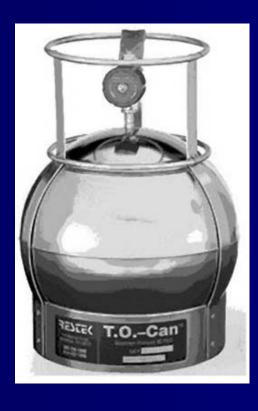


Passive samplers (Cont'd)





Passive/Active samplers - Canisters







- Canister Sampling Two basic modes
  - Grab sampling
    - Sample is taken over a short interval, i.e. 1 to 5 minutes
  - Integrated sampling
    - Sample is taken over an extended period,
      e.g. 0.5 to 2 hours or 0.5 to 24 hours

- Direct-reading Devices
- Two major types:
  - Colorimetric Devices
    - Concentration of VOCs indicated by color change
    - Examples: Dräger/indicator tube by active or passive sampling
  - Electronic Direct Reading Instruments
    - Comprised of detector, signal processing unit, data display and pump (no pump for passive instrument)
    - Examples: PID, FID, IR

Colorimetric Devices – Indicator Tube (Active)





- Indicator Tube Operations Step 1
  - Break tube tips, push in pump handle, and insert tube



- Indicator Tube Operations Step 2
  - Align guides, pull out handle until locked, and wait for flow finish indicator to pop out



- Indicator Tube Operations Step 3
  - Read measurement at end of color changed layer. Mark on the tube with pen



Colorimetric Devices – Indicator Tube (passive)





- Electronic Direct Reading Instruments
- Metal Oxide Semiconductor Sensor (MOS)
  - Change in resistance of a sensitive metal oxide layer induced by the surface interaction with ambient gases
  - Compact and low cost
  - Suffer from humidity sensitivity, non-linear response and long term drift
  - Response to inorganic gases, such as NO, NO2 or CO
  - Not suitable to measure low concentrations of VOCs where inorganic gases are present in higher concentrations

- Electronic Direct Reading Instruments (Cont'd)
- Metal Oxide Semiconductor Sensor (MOS) (Cont'd)





- Electronic Direct Reading Instruments (Cont'd)
- Photo-ionization Detector (PID)
  - Use an ultraviolet (UV) light source to ionize the VOCs
  - Detect or measure the charge of the ionized gas with the charge being a function of the concentration of VOCs
  - No change to the sample gas, that is non-destructive
  - Give a total value of VOCs, not individual VOCs
  - Sensitivity differ to different individual VOCs

- Electronic Direct Reading Instruments (Cont'd)
- Photo-ionization Detector (PID) (Cont'd)









- Electronic Direct Reading Instruments (Cont'd)
- Flame Ionization Detector (FID)
  - Use a flame as source to break down VOCs into ions
  - Detect or measure the charge of the ionized gas with the charge being a function of the concentration of VOCs
  - Change to the sample gas, that is destructive
  - Give a total value of VOCs, not individual VOCs
  - Sensitivity differ to different class VOCs
  - Require hydrogen as fuel therefore not portable and high running costs compare to other types of detector/sensor

- Electronic Direct Reading Instruments (Cont'd)
- Flame Ionization Detector (FID) (Cont'd)

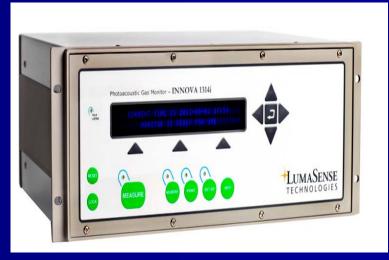




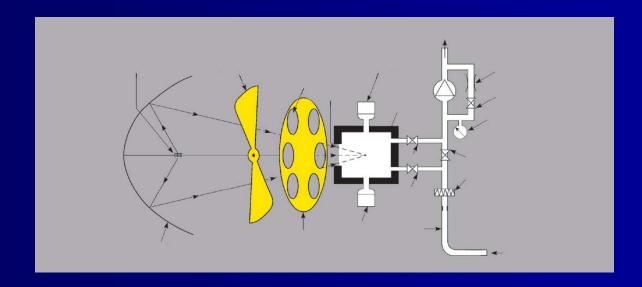
- Electronic Direct Reading Instruments (Cont'd)
- Photoacoustic Infrared Detector
  - Use Infrared (IR) (lamp) as source
  - Gas concentration is measured electro-optically by its absorption of specific wavelength in IR spectrum
  - Use optical filter to eliminate all light except the wavelength that the selected gas molecules can absorb
  - Gas molecules absorb some of the light energy and convert it into an acoustic signal which is detected by microphones
  - Can measure individual VOCs by selecting different filters
  - No change to the sample gas, that is non-destructive

- Electronic Direct Reading Instruments (Cont'd)
- Photoacoustic Infrared Detector (Cont'd)





- Electronic Direct Reading Instruments (Cont'd)
- Photoacoustic Infrared Detector (Cont'd)



#### Sample Analysis

#### Extraction

 Use known volume of solvent, e.g. carbon disulphide (CS<sub>2</sub>) to extract the target analytes from the samplers (active or passive), except canister and thermal desorption tubes (direct instrumentation analysis)

#### Instrumentation Analysis

Gas Chromatography coupled with flame ionization detector (GC-FID), Gas Chromatography coupled with mass selective detector (GC-MSD) and High performance liquid chromatography (HPLC)

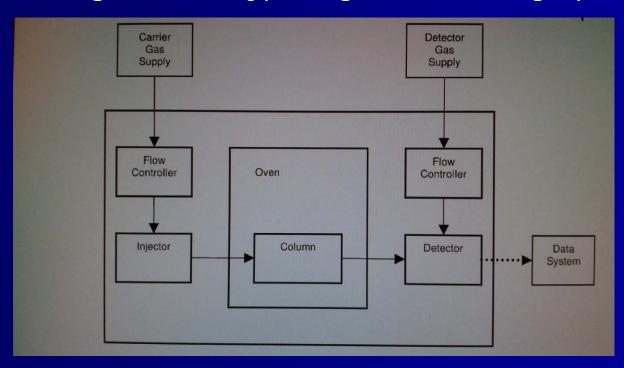
- Analytical Procedure Example
- NIOSH 1501
- Analyte: Benzene, Toluene, Ethyl Benzene, Xylenes
- Sampler: Solid Sorbent Tube (coconut shell charcoal)
- Flow rate: ≤ 0.20 L/min
- Volume: 5 to 30 L (benzene)
- Desorption: Carbon Disulphide (CS<sub>2</sub>)
- Detection: GC-FID

- Analytical Procedure Example (Cont'd)
- Sample Preparation Procedure
  - Place the front and back sorbent sections\* of the sampler tube in separate vials
  - Add 1.0 mL CS<sub>2</sub> to each vial
  - Attach crimp cap to each vial immediately
  - Allow to stand at least 30 min with occasional agitation
  - Ready for GC-FID analysis

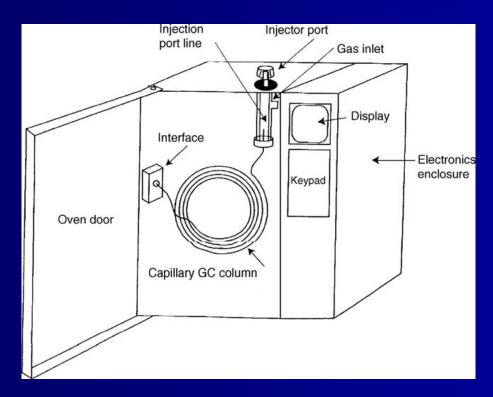
<sup>\*</sup> Used for checking breakthrough

- Instrumentation Analysis
- GC-FID or GC-MSD
  - Injection system
    - Hot or Cold
    - Split or Splitless
  - Oven
    - Isothermal or temperature programming
  - Column
    - Packed or capillary
  - Detector
    - FID, MSD or others

- Instrumentation Analysis (Cont'd)
- Block diagram of a typical gas chromatograph



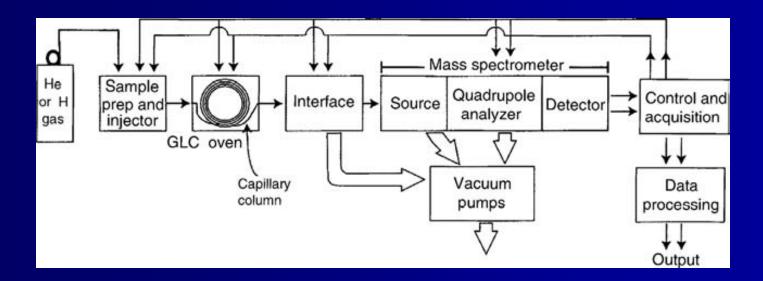
- Instrumentation Analysis (Cont'd)
- Typical GC System



- Instrumentation Analysis (Cont'd)
- Typical GC-FID System



- Instrumentation Analysis (Cont'd)
- Typical GC-MSD System

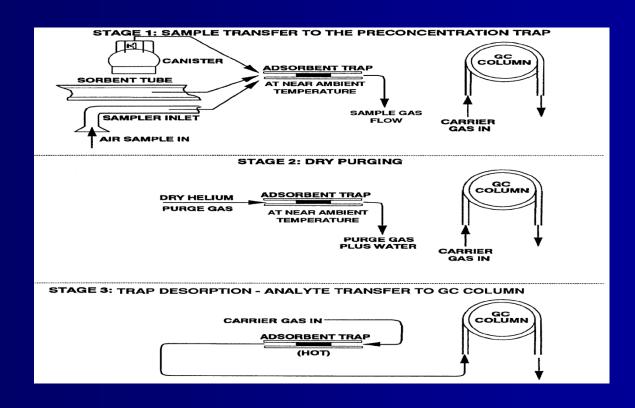


- Instrumentation Analysis (Cont'd)
- Typical GC-MSD System



- Instrumentation Analysis (Cont'd)
- Analytical System for Canister Sampling (Active/Passive)
- Preconcentrator
  - To remove water content and carbon dioxide
  - As an inlet to GC-MSD system
- GC-MSD System
  - To determine the VOCs concentration
  - Normally reported in ppbv level

- Instrumentation Analysis (Cont'd)
- Analytical System for Canister Sampling (Cont'd)



- Instrumentation Analysis (Cont'd)
- Analytical System for Canister Sampling (Cont'd)







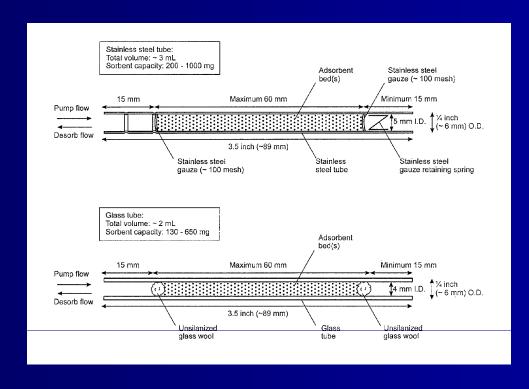
- Instrumentation Analysis (Cont'd)
- Analytical System for Canister Sampling (Cont'd)



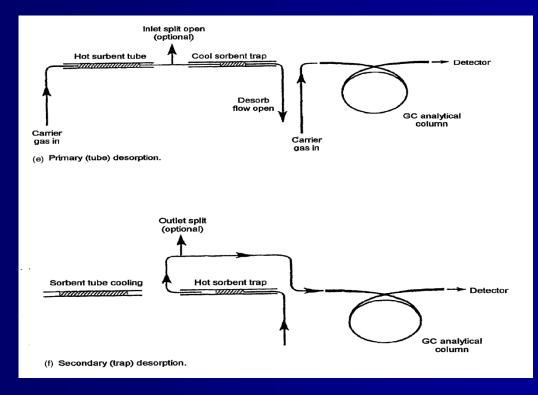


- Instrumentation Analysis (Cont'd)
- Analytical System for Thermo-desorption Tube Sampling
- Thermo-desorber
  - To remove air and refocuses analytes on a secondary trap
  - As an inlet to GC-MSD system
- GC-MSD System
  - To determine the VOCs concentration
  - Reported in ppmv or ppbv level

- Instrumentation Analysis (Cont'd)
- Analytical System for Tubes Sampling (Cont'd)



- Instrumentation Analysis (Cont'd)
- Analytical System for Tubes Sampling (Cont'd)



- Instrumentation Analysis (Cont'd)
- Analytical System for Tubes Sampling (Cont'd)





#### Accreditation

- What is accreditation?
- ISO/IEC 17000 definition
  - "Third party attestation related to a conformity assessment body conveying a formal demonstration of its competence to carry out specific conformity assessment tasks"
- What is the meaning?

- "Accreditation Body" as "authoritative body that performs accreditation"
- Hong Kong Accreditation Service (HKAS) is the "Accreditation Body" in Hong Kong
- HKAS under Innovation and Technology Commission (ITC) of HKSAR

- Conformity assessments activities are activities carried out to demonstrate that <u>specified requirements</u> relating to a product, process, system, person or body are fulfilled.
- Examples are testing, calibration, certification and inspection.

- A third party is independent of both the conformity assessment body to be accredited and its customers
- Accreditation is granted to conformity assessment bodies, i.e., organisations performing testing, calibration, certification, inspection, etc. rather than individuals

- "Competence" as "the ability to do something well or effectively"
- Most effort of accreditation is spent for assessing and monitoring competence.
- accreditation is applicable only to "specific conformity assessment tasks". In other words, accreditation is task specific.

- the specific tasks an accredited organisation is accredited for performing are activities listed in its scope of accreditation.
- Any activities not listed in the scope are not covered under the accreditation.

- Participation in HKAS accreditation schemes is voluntary
- In Hong Kong SAR, there is no statute to require laboratories, certification bodies or inspection bodies to obtain accreditation.

#### Benefits of Accreditation

- Through gaining accreditation, organisations obtain formal and third party recognition of their competence
- They can then demonstrate to their customers and to appropriate authorities that they are capable of working in accordance with currently accepted standards and practice

- Benefits of Accreditation (Cont'd)
  - both accredited organisations and their customers, and through them community at large, benefit from accreditation

# Thank you